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pHB log<sub>10</sub>aH where: pHB assigned pH of NIST buffer. The operational pH scale is used to measure sample pH and is defined as: pH<sub>x</sub> pHB F(Ex Es) 2.303 RT

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where:  $pH_x$  potentiometrically measured sample pH,  $F$  Faraday: 9.649 104 coulomb/mole,  $E_x$  sample emf,  $V$ ,  $E_s$  buffer emf,  $V$ ,  $R$  gas constant; 8.314 joule/(mole °K), and

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two major factors that influence

selection of the method to determine

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presence of interferences. In general, direct manual determination of low concentrations of ammonia is confined to drinking waters, clean surface

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