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a mixture in which a particles of a material are more or less evenly dispersed throughout a liquid or gas. colloid. a mixture consisting of tiny particles that are intermediate in size between those in solutions and those in suspensions and that are suspended in a liquid, solid, or gas.

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calculations that involve relating a solution's molar concentration and molal concentration will require information about the density of the solution mol solute kg solvent example: A solution is prepared by dissolving 13.5 g glucose (C 6H 12O 6) in 0.100 kg of water. Calculate the mass % glucose in this solution, and the molality of the solution.

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A solution that contains less solute than a saturated solution... Solution that contains more dissolved solute than a saturated... Physical state in which the opposing process of dissolution &... Chemistry Chapter 12. Any chemical that is dissolved in water. Intermediate size particles, heterogeneous, gelatin, paint,...

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